

## Module Descriptions

Module designation	Biochemistry <i>Biokimia</i>
Course Code	8420502047
Semester/Term	1 <sup>st</sup> Semester
Person responsible for the module	Prof. Dr.sc.agr. Yuni Sri Rahayu, M. Si. Prof. Dr. Isnawati, M. Si. Elma Sakinatus Sajidah, S.Si., M.Si., PhD
Language	Bahasa Indonesia (Indonesian language)
Relation to curriculum	Compulsory course
Teaching methods	Lecture
Workload	3 x 50 minutes lectures, 3 x 60 minutes structured activity, 3 x 60 minutes individual activity, 14 weeks per semester, 90 total hours per semester ~ 4.77 ECTS**
Credit Point	3 CU(4.77 ECTS)
Required and recommended prerequisites for joining the module	General Biology, General Chemistry
Module Objectives/intended learning outcomes	<p>After taking this course, students will be:</p> <ol style="list-style-type: none"> <li>1. Mastering concepts related to biochemical processes that occur in living things (<i>knowledge</i>)</li> <li>2. Apply biochemical concepts or theories that have been mastered in solving problems procedurally in accordance with their field of science (<i>knowledge</i>)</li> <li>3. Make the right decisions based on analysis of information and data, and be able to provide guidance in choosing various alternative solutions independently and groups in the field of biochemistry (<i>special competence</i>)</li> <li>4. Have a responsible, honest, careful, careful attitude in carrying out work practices related to the concept of biochemistry in the laboratory (<i>attitude</i>)</li> </ol>
Content	This course discusses organic compounds such as carbohydrates, proteins, lipids and nucleic acids and the metabolic processes (catabolism and anabolism) of these organic compounds in the body of living things, disorders that arise if there are abnormalities related to the supply and metabolism of these compounds. Other compounds which are important for life processes, such as water, vitamins and minerals are also discussed in this course. This course also

	<p>discusses enzymes which are important biological catalysts and hormones which are part of the coordination system of living things composed of chemical compounds. This subject also facilitates students to gain practical experience in identifying organic compounds. The attitude of students to be responsible, honest, work carefully, is carefully trained and developed in this course. This biochemistry course is packaged in the form of theory and practice to facilitate various process skills (minds on activity and hands on activity). It is hoped that students will be able to use their knowledge in the field of biochemistry to solve problems in daily life related to biochemistry. Learning is packaged with a student-centered learning approach using various methods according to the characteristics of the material, namely presentations, discussions, assignments, practicums and other learning strategies.</p>																																								
<p>Study and examination requirements and forms of examination</p>	<p><b>Study Requirement</b></p> <p>Attendance: students must attend at least 75% of the lectures to be eligible for the final examination.</p> <p><b>Study examination</b></p> <p>The final grade (<i>NA</i>) is calculated based on the following ratio:</p> <table border="1" data-bbox="632 1057 1224 1394"> <thead> <tr> <th data-bbox="632 1057 933 1147">Assessment Components</th><th data-bbox="933 1057 1224 1147">Percentage of contribution</th></tr> </thead> <tbody> <tr> <td data-bbox="632 1147 933 1192">Participation</td><td data-bbox="933 1147 1224 1192">20%</td></tr> <tr> <td data-bbox="632 1192 933 1236">Assignment</td><td data-bbox="933 1192 1224 1236">30%</td></tr> <tr> <td data-bbox="632 1236 933 1281">Mid-semester test</td><td data-bbox="933 1236 1224 1281">20%</td></tr> <tr> <td data-bbox="632 1281 933 1394">Final semester test</td><td data-bbox="933 1281 1224 1394">30%</td></tr> </tbody> </table> <p>Grade conversion of 0-100 scale into 0-4 scale is set as below:</p> <table border="1" data-bbox="636 1483 1430 1921"> <thead> <tr> <th data-bbox="636 1483 763 1528">Letter</th><th data-bbox="763 1483 1049 1528">Number</th><th data-bbox="1049 1483 1430 1528">Grade Interval</th></tr> </thead> <tbody> <tr> <td data-bbox="636 1528 763 1573">A</td><td data-bbox="763 1528 1049 1573">4,00</td><td data-bbox="1049 1528 1430 1573">85 ≤ A ≤ 100</td></tr> <tr> <td data-bbox="636 1573 763 1618">A-</td><td data-bbox="763 1573 1049 1618">3,75</td><td data-bbox="1049 1573 1430 1618">80 ≤ A- &lt; 85</td></tr> <tr> <td data-bbox="636 1618 763 1663">B+</td><td data-bbox="763 1618 1049 1663">3,50</td><td data-bbox="1049 1618 1430 1663">75 ≤ B+ &lt; 80</td></tr> <tr> <td data-bbox="636 1663 763 1708">B</td><td data-bbox="763 1663 1049 1708">3,00</td><td data-bbox="1049 1663 1430 1708">70 ≤ B &lt; 75</td></tr> <tr> <td data-bbox="636 1708 763 1753">B-</td><td data-bbox="763 1708 1049 1753">2,75</td><td data-bbox="1049 1708 1430 1753">65 ≤ B- &lt; 70</td></tr> <tr> <td data-bbox="636 1753 763 1798">C+</td><td data-bbox="763 1753 1049 1798">2,50</td><td data-bbox="1049 1753 1430 1798">60 ≤ C+ &lt; 65</td></tr> <tr> <td data-bbox="636 1798 763 1843">C</td><td data-bbox="763 1798 1049 1843">2,00</td><td data-bbox="1049 1798 1430 1843">55 ≤ C &lt; 60</td></tr> <tr> <td data-bbox="636 1843 763 1888">D</td><td data-bbox="763 1843 1049 1888">1,00</td><td data-bbox="1049 1843 1430 1888">40 ≤ D &lt; 55</td></tr> <tr> <td data-bbox="636 1888 763 1921">E</td><td data-bbox="763 1888 1049 1921">0,00</td><td data-bbox="1049 1888 1430 1921">0 ≤ E &lt; 40</td></tr> </tbody> </table>	Assessment Components	Percentage of contribution	Participation	20%	Assignment	30%	Mid-semester test	20%	Final semester test	30%	Letter	Number	Grade Interval	A	4,00	85 ≤ A ≤ 100	A-	3,75	80 ≤ A- < 85	B+	3,50	75 ≤ B+ < 80	B	3,00	70 ≤ B < 75	B-	2,75	65 ≤ B- < 70	C+	2,50	60 ≤ C+ < 65	C	2,00	55 ≤ C < 60	D	1,00	40 ≤ D < 55	E	0,00	0 ≤ E < 40
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<p>Reading List</p>	<p>1. Eubanks, P.L., Middlecamp, H.C., Pienta, J.N., Heltzel, E.C., Weaver, C.G. 2006. <i>Chemistry in Context</i>. New York: The McGraw-Hill Companies, Inc.</p>																																								

	<ul style="list-style-type: none"><li>2. Isnawati. 2010. <i>Biokimia</i>. Surabaya: Unesa Press.</li><li>3. Nelson, DL dan Cox, MM. 2015. <i>Principles of Biochemistry</i>. (Lehninger). New York: Worth Publishing Inc.</li><li>4. Rahayu, YS, Ratnasari, E., Isnawati. 2016. <i>Biochemistry</i>. Surabaya: Unesa Press.</li></ul>
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