MODUL HANDBOOK

Module Name	Mathematics For Chemistry		
Module level	Bachelor		
Abbreviation, if applicable	3074213022		
Sub-heading, if applicable	-		
Course included in the	-		
module, if applicable			
Semester/term	2 nd /First Year		
Module coordinator(s)	Dr. Nuniek Herdyastuti, M.Si		
Lecturer(s)	Dr. Pirim Setiarso, M.Si		
Language	Indonesian		
Classification within the curriculum	Compulsory Course		
Teaching format/class hours per week during the semester:	3 hours lecturers (50 min per hours)		
Workload:	3 x 50 minutes lectures, 3 x 60 minutes structured activity, 3 x 60 minutes individual activity, 14 weeks per semester, 119 total hours per semester ~ 4.77 ECTS**		
Credit points:	3 SCU (3 x 1.59 = 4.77 ECTS)		
Prerequisites course(s):	Basic mathematics		
Targeted learning outcomes:	CLO 1 Students have Capable to demonstrate knowledge related to theoretical concepts about structure, dynamics, and energy, as well as the basic principles of separation, analysis, synthesis and characterization of chemicals.		
Content:	Introduction: Briefly discuss the subject of mathematics for chemistry. Functions and Limits Concept of differential Calculus of differential Integral concept Integration methods Improper integrals Line integral and integral fold Operator Matrices: The definition of a matrix, matrix operations include addition of matrices, subtraction of matrices, multiplication of matrices and transpose matrix and properties as well as inverse matrix by Gauss substitution. Applied of matrices to solve problems in chemistry such as		

*1 CU in learning process = three periods consist of: (a scheduled instruction in a classroom or laboratory (50 minutes (b) structured activity (60 minutes); and (c) individual activit (60 minutes) according to the Regulation of Indonesia Ministr of Research, Technology, and Higher Education No. 44 Yea	Study / exam achievements:	reaction stoichiometry, redoxs reactions and quantitative analysis as well.Determinants include the definition of the matrix determinant and its properties as well as the minors and cofactors related to adjoint matrices and inverse matrices.Applied of the determinant matrices for quantitative analysis and determining the eigenvalues of the Schodinger equation phi electron energy in chemical compounds with double bondsDifferential equation: Definition of a differential equation, Differential equation with separate variables Homogeneous differential equations Exact Differential Equations Level 1 Linear Differential Equations Level n Linear Differential Equations Vector and tensor Sequence and series Special functions Fourier and Laplace transformsStudents are considered to complete the course and pass if they obtain at least 40% of maximum final grade. The final grade (NA) is calculated based on the following ratio:Assessment ComponentsPercentage of contribution Participation		
Media: Computer, LCD, White board Learning Methods Lectures and discussions, and working on problems Literature: 1. Robert G Mortimer, 2005, Mathematics for Physical Chemistry, 3th ed, Elsevier Inc, USA. 2. Irwin Krizig, 1989, Advanced Mathematic for Physicist and Engineering, 4thed, John Wiley & Sons Inc, New York. *1 CU in learning process = three periods consist of: (a scheduled instruction in a classroom or laboratory (50 minutes) (b) structured activity (60 minutes); and (c) individual activit (60 minutes) according to the Regulation of Indonesia Ministr of Research, Technology, and Higher Education No. 44 Yea 2015 jo. The Regulation of Indonesia Ministry of Research		Mid-semester test		
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**1 CU = 1,59 ECTS according to Rector Decree Of Universita	Notes:	scheduled instruction in a classroom or laboratory (50 minutes); (b) structured activity (60 minutes); and (c) individual activity (60 minutes) according to the Regulation of Indonesia Ministry of Research, Technology, and Higher Education No. 44 Year 2015 jo. The Regulation of Indonesia Ministry of Research,		