MODULE HANDBOOK

Module Name	Food Chemistry		
Module level	Bachelor		
Abbreviation, if applicable	3074113075		
Sub-heading, if applicable	-		
Course included in the	-		
module, if applicable			
Semester/term	6 /Third Year		
Module coordinator(s)	Dr. Prima Retno Wikandari, M.Si.		
Lecturer(s)	Prof. Dr. Lenny Yuanita, M.Si.		
	Mirwa Adiprahara, S.Si., M.Si.		
Language	Indonesian		
Classification within the	Elective Course		
Curriculum			
Teaching format/class	2 hours lecturers (50 min per hours)		
hours per week during the			
semester:	2 50 1		
Workload:	2×50 minutes lectures, 2×60 minutes structured activity,		
		nutes individual activity, 14 weeks per semester,	
Cradit points:		l hours per semester ~ 3.18 ECTS** 59 =3.18 ECTS	
Credit points:			
Prerequisite course(s): Targeted learning outcomes:	Biochemistry Structure and Function of BiomoleculesCLO 1.Students capable to demonstrate knowledge related		
Targeted learning outcomes.	CLU I.	to theoretical concepts about structure, composition,	
		and properties of food ingredient as well as the basic	
		principle of the chemical and physical changes of	
		food ingredients during processing and storage.	
	CLO 2.	Able to apply the knowledge in the field of food	
		chemistry obtained, and have the initiative in solving	
		community issues in the food sector	
	CLO 3.	Applying logical, critical, systematic and innovative	
		thinking in the context of development or	
		implementation of food science, that regards and	
		applies humanities in accordance with food	
	CLO 4.	chemistry in solving problems Capable to make decisions based on	
	CLU 4.	Capable to make decisions based on data/information in order to complete their	
		responsibility assignment and evaluate the	
		performance that has been done both individually	
		and in groups, have an entrepreneurial spirit with	
		environmental insight	
Content:	Introduction scope of food chemistry, food composition,		
	structure and properties, the positive and negative effect of food		
	processing, the types of food processing		
Structure and properties of food			

	salting out, salting in gelling, foaming, emuls b. structure of mono, di a FOS, inuline, solut emulsifier, stabilizer, th c. structure saturated and and unvisible fat, sapo (serebrosida, sfingomil boiling point, melting p emulsifier Functional foods : bioactive pe	and polisacharide, dietary fiber, bility, mutaroation, gelling, ickening, edible film unsaturated fatty acid, visible onified and unsaponified lipid ein, plasmogen, ester sterol), oint, cristal structure, plasticity, eptides, short chain fatty acids,		
	poliunsaturated fatty acids, antioxidant, FOS, inuline Changes during processing and storage a. Denaturation, hydrolisis, cross link, maillard reaction			
	peptide formation.			
	b. autooksidasi, hidrogenasi, trans fatty acid, ranciditas c. carbohidrat hydrolisis, dehidation, caramelisation,			
	maillard, swelling			
	Food additive			
	 a. Definition, types and funntion BTM b. Regulation Problems solving in mall nutrition and mall consumption a. Protein mall nutrition b. Effect of mall consumption of lipid and carbohydrate to hyperglycemia and hypercholesrerolemik c. Effect of trans fatty acids, free radical, autooxidation Propose a pilot project of functional food product as an 			
	entrepreunerally spirit	incuonal lood product as an		
Study / exam achievements:	Students are considered to complete the course and pass if they			
	obtain at least 40% of maximum final grade. The final grade			
	(NA) is calculated based on the following ratio:			
	Assessment Components	Percentage of contribution		
	Participation	20%		
	Assignment	30%		
	Mid-semester test	20%		
	Final semester test	30%		
Media:	Computer, LCD, White board			
Learning Methods	Individuals assignment, group assignment, discussion,			
Literature:	presentation. de Man, John M. 1990. Principle of Food Chemistry, 2 nd ed. An AV1 book, Van Nostrang aReinhold, New York Fennemas, Food Chemistry 2007. 4 th Edition, edited by Srinivasan Damodaran, CRC Press.			

	*1 CU in learning process = three periods consist of: (a)
Notes:	scheduled instruction in a classroom or laboratory (50
	minutes); (b) structured activity (60 minutes); and (c)
	individual activity (60 minutes) according to the Regulation of
	Indonesia Ministry of Research, Technology, and Higher
	Education No. 44 Year 2015 jo. the Regulation of Indonesia
	Ministry of Research, Technology, and Higher Education No.
	50 Year 2018.
	**1 CU = 1,59 ECTS according to Rector Decree Of
	Universitas Negeri Surabaya No. 598/UN38/Hk/Ak/2019