MODULE HANDBOOK

Modul Nama	Capita Salaata		
Module Level	Bachelor of Chemistry		
Abbraviation if	Bachelor of Chemistry		
applicable	5074112008		
Sub-heading, if applicable	-		
Course included in the	-		
module, if applicable			
Semester/term	7 th / Fourth Year		
Modul coordinator(s)	Prof. Dr. Suyatno, M.Si. (C1), J Cahyaningrum, M,Si. (C2), Dr. Gusti Made Sanjaya, M.Si. (C4 M.Si. (C4)	Prof. Dr. Sari Edi Pirim Setiarso, M.S. (C3), Dr. I), Dr. Nuniek Herdyastuti,	
Lecturer(s)	Prof. Dr. Suyatno, M.Si. (C1),	Prof. Dr. Sari Edi	
	Cahyaningrum, M,Si. (C2), Dr. Gusti Made Sanjaya, M.Si. (C4 M.Si. (C4)	Pirim Setiarso, M.S. (C3), Dr. I), Dr. Nuniek Herdyastuti,	
Language	Indonesian Language		
Classification within the curriculum	Elective Course		
Teaching format/class hours	2 hours lectures (50 min / hour)		
per week during			
the semester			
Workload	2×50 minutes lectures, 2×60 minutes lectures 2×60 minutes 2×60 minutes lectures 2×60 minutes 2×60 min	minutes structured activity,	
	2 x 60 minutes individual activi	ty, 14 weeks per semester,	
	79,33 total hours per semester ~	- 3.18 ECTS**	
Credit point	2 CU x 1.59 = 3.18 ECTS		
Requirement	-		
Learning Outcomes	General Competence (knowledge):		
	Student can conclude recent development in analytical		
	chemistry, physical chemistry	, organic chemistry, inorganic	
	chemistry, and bochemistry		
	Spesific Competence :		
	At the end of the lecture, students can conclude recent		
	development in analytical chemistry, physical chemistry,		
Contant	Course materials discuss th	he understanding of recent	
Content	development in analytical of	he understanding of recent	
	organia chemistry inorgania che	memistry, physical chemistry,	
Study/avam achievements	Students are considered to com	plate the course and pass if they	
Study/exam acmevements	obtain at least 40% of maximum final grade. The final grade		
	(NA) is calculated based on the following ratio:		
	A appagement Community	Demoente og of contribution	
	Assessment Components	reicentage of contribution	
	Participation	20%	
	Assignment	30%	
	Mid-semester test	20%	
	Final semester test	30%	

Forms of media	Computer, LCD, White board	
Learning Methods	Lectures, discussion, problem solving, assignment	
Literatur	1. Abdullah, M. (2009). Pengantar Nanosains. Bandung: ITB	
	2. Abdullah, M. (2010). Karakterisasi Nanomaterial. Bandung: ITB	
	3. Dewick, P. M. (2009). <i>Medicinal Natural Products</i> . 3 rd Edition. New	
	York: John Wiley and Sons, Inc.	
	4. Glik, B.R. and Pasternak, J.J. (1994). Molecular Biotechnology:	
	Principles and Application of Recombinant DNA. Washington, D.C.:	
	ASM Press.	
	5. Hofmann, A. (2018). <i>Physical Chemistry Essentials</i> . Australia:	
	Springer.	
	6. Karlin, K.D. (2003). Progress in Inorganic Chemistry. Vol 51. New	
	Jersey: John Wiley and Sons, Inc.	
	/. Nelson, D.L. and Cox, M.M. (2003). Lehninger: Principles of	
	Biochemistry. 4 rd Edition. University of Winconsin Madison	
	8. Pearce, E.M., et al. (2015). Physical Chemistry Research for	
	Engineering and Applied Sciences. Vol 3. Canada: Apple Academic	
	Press, Inc. O Wong I (1004) Analytical Electrophemiatry, New York, VCII	
	Publisher.	
	10. Jurnal terkini bidang Kimia Analitik, Kimia Organik, Kimia	
	Anorganik, Kimia Fisika, dan Biokimia	
	*1 CU in learning process = three periods consist of: (a) scheduled	
	instruction in a classroom or laboratory (50 minutes); (b) structured	
	activity (60 minutes); and (c) individual activity (60 minutes) according	
	to the Regulation of Indonesia Ministry of Research, Technology, and	
Notes:	Higher Education No. 44 Year 2015 jo. The Regulation of Indonesia	
	Ministry of Research, Technology, and Higher Education No. 50 Year	
	2018.	
	**1 CU = 1,59 ECTS according to Rector Decree Of Universitas Negeri	
	Surabaya No. 598/Un38/Hk/Ak/2019	