MODULE HANDBOOK

Module Name	Environmental Chemistry	
Module level	Bachelor	
Abbreviation, if applicable	8420403154	
Sub-heading, if applicable	-	
Course included in the	-	
module, if applicable		
Semester/term	6 th /Third Year	
Module coordinator(s)	Prof. Dr. Suyono, M.Pd	
Lecturer(s)	Prof. Dr. Suyono, M.Pd, Dr. Amaria, M.Si. Rusmini S.Pd,	
	M.Si, Dina Kartika Maharani S.Si, M.Sc	
Language	Bahasa Indonesia	
Classification within the	Elective Course	
curriculum		
Teaching format/class	3 hours lectures (50 min / hour)	
hours per week during the		
semester:		
Workload:	3 x 50 minutes lectures, 3 x 60 minutes structured activity,	
	3 x 60 minutes individual activity, 14 weeks per semester,	
	119 total hours per semester ~ 4.77 ECTS**	
Credit points:	3 CU = 3 x 1.59 = 4.77 ECTS	
Prerequisite course(s):	-	
Targeted learning outcomes:	 Students have knowledge about the sources, reactions, displacement, effects, and changes of chemical species in air, water and soil, the reciprocal effect of human activities on all those mentioned, and an analysis of environmental impacts (Amdal) Students are skilled at using tools in experimenting with water quality parameters from the environment Students have the ability to work together and are responsible for discussing knowledge about 1) sources, reactions, displacement, effects, and changes in chemical species in air, water and soil, 2) The reciprocal effect of human activities on all the so-called on no.1 and 3) Environmental impact analysis (Amdal) Students have the ability to communicate knowledge about 1) sources, reactions, displacement, effects, and changes in chemical species in air, water and soil, 2) The reciprocal effect of human activities on all the so-called on no.1 and 3) Environmental impact analysis (Amdal) 	
Content:	 Sources, reactions, displacement, effects, and changes in chemical species in air, water and soil, 	
	 The reciprocal effect of human activities on all the so- called on no.1 and 3) 	
	3. Environmental impact analysis (Amdal)	
Study / exam achievements:	Students are considered to complete the course and pass if	

	they obtain at least 40% of maximum final grade. The final grade (NA) is calculated based on the following ratio:		
	Assessment Components	Percentage of contribution	
	Participation	20%	
	Assignment	30%	
	Mid-semester test	20%	
	Final semester test	30%	
Media:	Computer, LCD, White board, laboratory		
Learning Methods	Individuals assignment, group assignment, discussion, presentation, and practicum		
Literature:	 De, anil Kumar. 1987. Environmental Chemistry. India: Willey Eastern Limited. 2.Faust, S.D and Aly, O.M.1981. Chemistry of Natural Water. London: Ann Arbor Science. 		
	3.Manahan, S.E. 1994. <i>Environmental Chemistry</i> . London: Lewis Publishers CRC Pres.Inc		
	4.More, J.W. and More, E.A., 1976. <i>Environmental Chemistry</i> . New York: Academic Press.		
	5.Radojevic, Miroslav and Bashkin, Vladimir N, 1999, <i>Practical Environmental Analysis</i> , Cambridge : Royal Society of Chemistry		
Notes:	 *1 CU in learning process = three periods consist of: (a) scheduled instruction in a classroom or laboratory (50 minutes); (b) structured activity (60 minutes); and (c) individual activity (60 minutes) according to the Regulation of Indonesia Ministry of Research, Technology, and Higher Education No. 44 Year 2015 jo. the Regulation of Indonesia Ministry of Research, Technology, and Higher Education No. 50 Year 2018. **1 CU = 1,59 ECTS according to Rector Decree Of Universitas Negeri Surabaya No. 598/Un38/Hk/Ak/2019 		