MODULE HANDBOOK

Module Name	Food Chemistry		
Module level	Bachelor		
Abbreviation, if applicable			
Sub-heading, if applicable	8420402167		
Course included in the	-		
	-		
module, if applicable			
Semester/term	7 th /Fourth Year		
Module coordinator(s)	Dr. Prima Retno Wikandari,M.Si		
Lecturer(s)	Prof. Dr. Lenny Yuanita, M.Si; Mirwa Adiprahara, S.Si., M.Si		
Language	Indonesian		
Classification within the	Elective Course		
Curriculum			
Teaching format/class	2 hours lecturers (50 min per hours)		
hours per week during the			
semester:			
Workload:	2 x 50 minutes lectures, 2 x 60 minutes structured activity,		
	2 x 60 minutes individual activity, 14 weeks per semester,		
	79,33 total hours per semester ~ 3.18 ECTS**		
Credit points:	2 CU = 2 x 1.59 = 3.18 ECTS		
Prerequisite course(s):	Biochemistry (Structure and Function of Biomolecules)		
Targeted learning outcomes:	 Students capable to demonstrate knowledge related to theoretical concepts about structure, composition, and properties of food ingredient as well as the basic principle of the chemical and physical changes of food ingredients during processing and storage. Able to apply the knowledge obtained in the field of food chemistry, and have the initiative to resolve public issues in the food sector. Applying logical, critical, systematic and innovative thinking in the context of development or implementation of food science, that regards and applies humanities in accordance with food chemistry in solving problems Capable to make decisions based on data/information in order to complete their responsibility assignment and evaluate the performance that has been done both individually and in groups, have an entrepreneurial spirit with environmental insight 		
Content:	 Introduction scope of food chemistry, food composition, structure and properties, the positive and negative effect of food processing, the types of food processing Structure and properties of food a. structure of amino acid, peptide and protein, amfoter, salting out, salting in, protein solubility ,swelling, gelling, foaming, emulsifier b. structure of mono, di and polisacharide, dietary fiber, 		
	b. structure of mono, di and polisacharide, dietary fiber FOS, inuline, solubility, mutaroation, gelling emulsifier, stabilizer, thickening, edible film		

	 c. structure saturated and unsaturated fatty acid, visible and unvisible fat, saponified and unsaponified lipid (serebrosida, sfingomilein, plasmogen, ester sterol), boiling point, melting point, cristal structure, plasticity, emulsifier Functional foods: bioactive peptides, short chain fatty acids, 		
	poliunsaturated fatty acids, antioxidant, FOS, inuline		
	Changes during processing and storage		
	a. Denaturation. , hydrolisis, cross link, maillard reaction peptide formation.		
	b. autooidation, hydrogenation, trans fatty acid, ranciditas		
	c. carbohidrat hydrolisis, dehidation, caramelisation,		
	maillard, swelling Food additive a. Definition, types and function BTM b. Regulation Problems solving in mall nutrition and mall consumption		
	a. Protein mall nutrition b. Effect of mall consumption of lipid and carbohydrate to		
	b. Effect of mall consumption of lipid and carbohydrate to hyperglycemia and hypercholesrerolemik		
	c. Effect of trans fatty acids, free radical, autooxidation		
		inctional food product as an	
	entrepreunerally spirit		
Study / exam achievements:	Students are considered to complete the course and pass if they obtain at least 40% of maximum final grade. The final grade (NA) is calculated based on the following ratio:		
	Assessment Components	Percentage of contribution	
	Participation	20%	
	Assignment	30%	
	Mid-semester test	20%	
	Final semester test	30%	
Media:	Computer, LCD, White board		
Learning Methods	Individuals assignment, group assignment, discussion,		
Literature:	 presentation. Belitz, H-D., Grosch,W., Schieberle, P., 2009. Food Chemistry, 4 threvised and extended ed.Springer-Verlag Berlin Heidelberg Schaschke, C.J., 2011. Food Processing. Ventus Publishing Aps 		
	Fennemas, Food Chemistry 2007. 4 th Edition, edited by Srinivasan Damodaran, CRC Press.		
	*1 CU in learning process = three periods consist of: (
	scheduled instruction in a	-	
Notes:	minutes); (b) structured activities	• • • • • • • • • • • • • • • • • • • •	
	individual activity (60 minutes) Indonesia Ministry of Resear	0 0	

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Ministry of Research, Technology, and Higher Education No.	
50 Year 2018.	
**1 CU = 1,59 ECTS according to Rector Decree Of	
Universitas Negeri Surabaya No. 598/Un38/Hk/Ak/2019	