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MINISTRY OF EDUCATION AND CULTURE

UNIVERSITAS NEGERI SURABAYA FACULTY OF MATHEMATICS AND NATURAL SCIENCES DEPARTMENT OF NATURAL SCIENCES

Ketintang Campus, Jl. Ketintang C12 Building, Surabaya 60231 Phone (031)18296427

Website http://pendidikan-sains.fmipa.unesa.ac.id

Undergraduate Programme in Science Education

Module Handbook

Module Name:	Kimia Rumah Tangga		
	(Household Chemistry)		
Module Level:	Bachelor degree/Undergraduate Programme		
Course Code:	8420102073		
Abbreviation, if applicable:	KRT		
Courses included in the	Not applicable		
module, if applicable:			
Semester/term	Elective		
Module coordinator(s):	Siti Nurul Hidayati, S.Pd., M.Pd.		
Lecturer(s):	Siti Nurul Hidayati, S.Pd., M.Pd. Wahyu Sabtiawan, S.Si., M.Pd		
Language:	Bahasa Indonesia (Indonesian Language)		
Classification within the curriculum:	Elective		
Teaching format/class hours	2 contact hours of lectures (Indonesia credit semester or sks*)		
per week during the			
semester:			
Workload:	2×50 minutes lectures, 2×60 minutes structured activity, 2×60		
	minutes individual activity, 14 weeks per semester, 79.33 total		
	hours per semester ~ 3.18 ECTS**		
Credit point:	2 sks (3.18 ECTS)		
Requirements:	Solution, Introduction to Biochemistry		
Learning	Course Learning Outcomes (CLOs):		
goals/competencies:	1. Able to take advantage of science and technology in studying matters related to household chemicals, additives in food, addictive substances (psychotropic) and able to		
	adapt to situations faced in solving problems.		
	2. Mastering the theoretical concepts of household chemicals,		
	additives in food, addictive substances (psychotropics) in		
	depth and formulating them in procedural problem solving.		
	3. Able to make decisions based on analysis of information and		
	data and provide guidance in choosing alternative solutions.		
	4. Responsible for informing the results of analysis of		
	information and data both orally and in writing.		
Content:	Scientific Method, Matter and Properties of Matter, Periodic		
	System of Elements, Chemical Bonds, Stoichiometry, Solutions,		
	Colloid Systems,		
Attribute Soft skill:	Discipline, collaboration, responsibility, and argumentation in the natural classroom setting		
Study/exam achievements:	Students are considered to be competent and pass if at least get		
ctay, exam demovements.	40% of the maximum final grade. The final grade (NA) is calculated		
	based on the following weight:		
	and the following weight.		
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	Assessment Components	Percentage Contribution		
	Participation	20%		
	Assignment	30%		
	Mid-semester test	20%		
	Final semester test	30%		
	Total	100%		
Learning Methods	deductive learning, lecturing	,		
	discussion, and presentation	(structured activities), and fl	ip	
	learning			
Form of Media:	LCD, PowerPoint, hand out, simulation, and whiteboard and e-			
	learning unesa			
	(https://vinesa.unesa.ac.id/course/view.php?id=423)			
Literature (primary	1. Helmprecht. H.L. and Friedman. L.T. 1997. basic Chemistry for			
references):	The Life Sciences. New York: Mc Graw Book Company2. Pusat Perbukuan. 2003. Ensiklopedia Sains dan Kehidupat Jakarta: Depdiknas			
	3. Lucy T Pride. 2010. Env	•	toduction.	
Cumming publishing company.				
Notes:	*1 sks in learning process = t	hree contact hours that con	sist of:	
	(a) scheduled instruction in a	a classroom or laboratory (50	0	
	minutes); (b) structured acti	vity (60 minutes); and (c) inc	lividual	
	activity (60 minutes) accordi	ng to the Regulation of Indor	nesia	
	Ministry of Research, Techno	logy, and Higher Education N	lo. 44	
	Year 2015 jo. the Regulation	of Indonesia Ministry of Rese	earch,	
	Technology, and Higher Educ	ation No. 50 Year 2018.		
	**1 sks = 1,59 ECTS			